# Indole Alkaloids from the Tunicate Aplidium meridianum 

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Five new indole alkaloids, meridianins A-E (1-5), have been isolated from the tunicate Aplidium meridianum, which was collected at a depth of 100 m near the South Georgia I slands, and their structures were elucidated by spectroscopic techniques. Compounds 2-5 showed cytotoxicity toward murine tumor cell lines.

Marine invertebrates are a very important source of antitumor secondary metabolites. Among these, 3-substituted indoles have frequently been isolated, especially from tunicates and sponges. The substituent at position 3 of indoles is often an additional heterocyclic ring, ${ }^{1-5}$ with recent examples including the didemnimides from the tunicate Didemnum conchyliatum, ${ }^{6}$ alboinon from Dendrodoa grossularia, ${ }^{7}$ and the sponge metabolites psammopemmins A-C (6-8). ${ }^{8}$ As part of our ongoing study of bioactive compounds isolated from South Atlantic invertebrates, we have examined the constituents of the tunicate Aplidium meridianum (Sluiter, $1906^{9,10}$ ) (family Polyclinidae) collected near the South Georgia Islands. In this paper we report the isolation and structure el ucidation of five novel indole alkaloids, meridianins A-E (1-5). All these compounds have a brominated and/or hydroxylated indole nucleus with a 2-aminopyrimidine substitutent at C-3.

TheHRMS of compound $\mathbf{1}$ gave the molecular formula $\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{4} \mathrm{O}$, indicative of 10 degrees of unsaturation, while the NMR data suggested the presence of two heteroaromatic moieties. The ${ }^{1} \mathrm{H}$ NMR signals (Table $1)$ at $\delta 8.20(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=1.2 \mathrm{~Hz})$ and $\delta 11.71(\mathrm{br} \mathrm{s}, 1 \mathrm{H})$, together with the UV spectrum of 1, indicated the presence of a 3-substituted indole. The characteristic ${ }^{1} \mathrm{H}$ NMR coupling pattern [ $\delta 6.36$ (dd, $1 \mathrm{H}, \mathrm{J}=7.1,0.7$ $\mathrm{Hz}), \delta 6.78(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=7.5,0.7 \mathrm{~Hz})$ and $\delta 6.96(\mathrm{dd}, 1 \mathrm{H}$, $\mathrm{J}=7.1,7.5 \mathrm{~Hz}$ )] suggested the presence of a hydroxyl group either at C-4 or C-7. A sharp downfield signal at $\delta 13.55\left(\mathrm{~s}, 1 \mathrm{H}\right.$, exchangeable with $\left.\mathrm{D}_{2} \mathrm{O}\right)$ confirmed the presence of a deshielded hydroxyl group. This was consistent with substitution at C-4, which was further supported by correlations observed in a COLOC spectrum and comparison with reference data. ${ }^{8}$ With this substructure established, we could then determine by subtraction the formula of the fragment at C-3 as $\mathrm{C}_{4} \mathrm{H}_{4} \mathrm{~N}_{3}$. This formula, together with ${ }^{1} \mathrm{H} N M R$ signals at $\delta 7.09(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=5.4 \mathrm{~Hz}), \delta 8.10(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=5.4 \mathrm{~Hz})$, and $\delta 6.69\left(\mathrm{~s}, 2 \mathrm{H}\right.$, exchangeable with $\left.\mathrm{D}_{2} \mathrm{O}\right)$, suggested the presence of either a 4'-substituted 2-aminopyrimi-

[^0]
$1 \mathrm{R}^{1}=\mathrm{OH} ; \mathrm{R}^{2}=\mathrm{R}^{3}=\mathrm{R}^{4}=\mathrm{H}$ 2 R1 $=\mathrm{OH} ; \mathrm{R}^{2}=\mathrm{R} 4=\mathrm{H} ; \mathrm{R} 3=\mathrm{Br}$ 3 R1=R3=R4=H; R2=Br $4 \mathrm{R}^{1}=\mathrm{R} 2=\mathrm{R} 4=\mathrm{H} ; \mathrm{R} 3=\mathrm{Br}$
$5 \mathrm{R}^{1}=\mathrm{OH} ; \mathrm{R}^{2}=\mathrm{R} 3=\mathrm{H} ; \mathrm{R}^{\mathbf{4}}=\mathrm{Br}$

$6 \mathrm{R}_{1}=\mathrm{H} ; \mathrm{R}^{2}=\mathrm{H}$
7 R1=H; R2=Br
$8 \mathrm{R}_{1}=\mathrm{Br} ; \mathrm{R}_{2}=\mathrm{H}$
dine or 2-aminopyridazine ring. The ${ }^{13} \mathrm{C}$ NMR downfield signals (Table 2) at $\delta 161.9$ (C), $\delta 160.6$ (C), and $\delta$ 158.5 (CH) favored the former structure. This was confirmed by interpetration of the COLOC spectrum and comparison with literature data. ${ }^{11}$ Assignment of all protonated carbons was obtained by selective heteronuclear decoupling experiments. Unambiguous assignment of the C-2'/C-4'and C-3/C-3a pairs was achieved through a gated decoupling experiment (DMSO-d 6 $-\mathrm{D}_{2} \mathrm{O}$ ). In this spectrum, $\mathrm{C}-2^{\prime}$ appeared as a clean doublet ( $\mathrm{J}=12.2 \mathrm{~Hz}$ ) due to the large $\mathrm{H}-6^{\prime} / \mathrm{C}-2^{\prime}{ }^{3} \mathrm{~J}$ coupling, resulting from a nitrogen atom in the coupling path. On the other hand, the signal for C-4'consisted of a doublet of doublets ( $\mathrm{J}=4.0 ; 2.7 \mathrm{~Hz}$ ) due to the smaller ${ }^{2} \mathrm{~J}$ and ${ }^{3} \mathrm{~J}$ couplings with $\mathrm{H}-5^{\prime}$ and $\mathrm{H}-6^{\prime} . .^{12}$ In the case of the C-3/C-3a pair, the C-3a multiplet showed three large ${ }^{3} \mathrm{~J}$ couplings with $\mathrm{H}-2, \mathrm{H}-5$, and $\mathrm{H}-7$, while C-3 appeared as a doublet due to the large ${ }^{2} \mathrm{~J}$ ( 8 Hz ) coupling with H-2 typical of indoles.

The HRMS of compounds 2-5 contained the correct clusters of peaks for the molecular formulas $\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{~N}_{4}-$ $\mathrm{OBr}, \mathrm{C}_{12} \mathrm{H}_{9} \mathrm{~N}_{4} \mathrm{Br}$, and $\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{~N}_{4} \mathrm{Br}, \mathrm{C}_{12} \mathrm{H}_{9} \mathrm{~N}_{4} \mathrm{OBr}$, respectively, indicating that all of these compounds also have 10 degrees of unsaturation. The NMR data of 2-5

Table 1. ${ }^{1} \mathrm{H}$ NMR Spectral Data $\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO}, 200 \mathrm{MHz}\right]$ for Compounds $1-5^{\text {a }}$

| proton | compound |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  | 1 | 2 | 3 | 4 | 5 |
| N1-H | 11.71 (brs) | 11.99 (s) | 11.85 (brs) | 11.76 (brs) | 11.86 (brs) |
| 2 | 8.20 (d; 1.2) | 8.32 (d; 2.8) | 8.24 (d; 2.5) | 8.21 (d; 2.6) | 8.29 (brs) |
| 4 |  |  | 8.75 (brd; 1.8) | 8.55 (d; 8.4) |  |
| 4-OH | 13.55 (s) | 13.92 (s) |  |  | 13.89 (s) |
| 5 | 6.36 (dd; 7.1, 0.7) | 6.54 (d; 1.5) |  | 7.24 (dd; 8.4, 1.8) | 6.37 (d; 8.3) |
| 6 | 6.96 (dd; 7.1, 7.5) |  | 7.28 (dd; 8.4, 1.8) |  | 7.19 (d; 8.3) |
| 7 | 6.78 (dd; 7.5, 0.7) | 7.02 (d; 1.5) | 7.41 (d; 8.4) | 7.63 (d; 1.8) |  |
| $2^{\prime}-\mathrm{NH}_{2}$ | 6.69 (s; 2H) | 7.02 (s; 2H) | 6.48 (s; 2H) | 6.43 (s; 2H) | 6.79 (s; 2H) |
| 5 ' | 7.09 (d; 5.4) | 7.18 (d; 5.5) | 7.00 (d; 5.5) | 7.00 (d; 5.1) | 7.23 (d; 5.4) |
| $6^{\prime}$ | 8.10 (d; 5.4) | 8.17 (d; 5.5) | 8.11 (d; 5.5) | 8.12 (d; 5.1) | 8.19 (d; 5.4) |

a $\delta$ in ppm, J in Hz.

Table 2. ${ }^{13} \mathrm{C}$ NMR Spectral Data for Compounds 1-5 [ $\delta$ in ppm, $\left.\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO}\right]$

|  | compound |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| carbon | $\mathbf{1}$ | $\mathbf{2}$ | $\mathbf{3}$ | $\mathbf{4}$ | $\mathbf{5}$ |
| 2 | 128.5 | 129.9 | 129.6 | 129.2 | 129.2 |
| 3 | 113.8 | 113.7 | 113.3 | 114.8 | 116.1 |
| 3 a | 114.5 | 114.0 | 127.1 | 124.5 | 115.2 |
| 4 | 152.1 | 153.0 | 124.6 | 124.3 | 152.0 |
| 5 | 105.6 | 108.8 | 113.4 | 123.1 | 107.3 |
| 6 | 124.4 | 116.7 | 124.7 | 113.9 | 126.7 |
| 7 | 102.4 | 105.3 | 113.9 | 114.5 | 92.6 |
| $7 a$ | 139.4 | 139.7 | 135.9 | 138.0 | 136.9 |
| $2^{\prime}$ | 161.9 | $160.7^{\text {a }}$ | $163.6^{\text {b }}$ | $163.6^{c}$ | $160.2^{\text {d }}$ |
| $4^{\prime}$ | 160.6 | $160.8^{\text {a }}$ | $162.3^{\text {b }}$ | $162.3^{c}$ | $161.8^{\text {d }}$ |
| $5^{\prime}$ | 104.5 | 104.6 | 105.4 | 105.4 | 104.8 |
| $6^{\prime}$ | 158.5 | 157.1 | 157.2 | 157.2 | 159.0 |
| a-d Values marked | with the sameletter may beinterchangeable |  |  |  |  |

${ }^{\text {a-d }}$ Values marked with the same letter may be interchangeable.
indicated the presence of a 2-aminopyrimidine ring similar to that of $\mathbf{1}$, while the coupling patterns of the indole protons clearly established their structural differences. The molecular formula of $\mathbf{2}$ indicated the presence of bromine and hydroxyl substituents in the indole ring. The ${ }^{1} \mathrm{H}$ NMR spectrum of $2[\delta 6.54(\mathrm{~d}, \mathrm{~J}=$ $1.5 \mathrm{~Hz}), \delta 7.02(\mathrm{~d}, \mathrm{~J}=1.5 \mathrm{~Hz})$ ] showed that the indole protons were meta coupled, while their upfield chemi cal shifts were in accordance with the presence of a hydroxyl substituent. A sharp downfield signal at $\delta 13.92$ ( $\mathrm{s}, 1 \mathrm{H}$, exchangeable with $\mathrm{D}_{2} \mathrm{O}$ ) located the hydroxyl group at C-4. The chemical shifts of all the carbon atoms of $\mathbf{2}$ were assigned from a COLOC experiment and comparison with literature data. ${ }^{8}$

The typical ${ }^{1} \mathrm{H}$ NMR AMX coupling pattern of $\mathbf{3}[\delta$ 7.28 (dd, J $=8.4,1.8 \mathrm{~Hz}$ ), $\delta 7.41(\mathrm{~d}, \mathrm{~J}=8.4 \mathrm{~Hz}$ ), $\delta 8.75$ (br d, J $=1.8 \mathrm{~Hz}$ )] placed the bromine substituent either at C-5 or C-6. However, the downfield shift of the H-4 signal, which appeared as a meta doublet, unequivocally positioned the bromine at C-5.

HRMS indicated that $\mathbf{3}$ and $\mathbf{4}$ were isomers. The ${ }^{1} \mathrm{H}$ NMR spectrum of $\mathbf{4}$ showed an AMX system with the downfield H-4 signal as an ortho doublet ( $J=8.4 \mathrm{~Hz}$ ), thus establishing that the bromine substituent was located at C-6.

The mol ecular formula of 5 indicated the presence of bromine and hydroxyl substituents in the indole ring. A downfield signal at $\delta 13.89$ (sharp s), together with two ortho-coupled protons [ $\delta 6.37(\mathrm{~d}, \mathrm{~J}=8.3 \mathrm{~Hz}$ ) and $\delta$ $7.19(\mathrm{~d}, \mathrm{~J}=8.3 \mathrm{~Hz})$ ], suggested the presence of a hydroxyl group at C-4 and a bromine either at C-7 or $\mathrm{C}-5$. The presence of NOE correlations in a phasesensitive NOESY experiment between $4-\mathrm{OH} / \mathrm{H}-5^{\prime}$ and 4-OH/H-5 confirmed that the bromine group was at C-7.

Assignment of the ${ }^{13} \mathrm{C}$ NMR chemical shifts of compound 5 (see Table 2) was completed with the help of COLOC, DEPT, and heteronuclear irradiation spectra.
Meridianins A (1), B (2), and E (5) and psammopemmins A-C (6-8) ${ }^{8}$ have a similar indole nudleus substitution pattern, but the latter compounds have a $5^{\prime}$ substituted 4 '-amino $2^{\prime}$-bromopyrimidine moiety as the additional heterocycle at $\mathrm{C}-3$. On the other hand, variolin B (9) isolated from the sponge Kirkpatrickia varialosa ${ }^{11}$ has a 2 -aminopyrimidine ring on a 3 -substituted azaindole, but differs from the meridianins in the presence of an additional heterocyclic ring. Compounds 2-5 showed cytotoxicity toward LMM3 (murine mamarian adenocarcinoma cell line) with $\mathrm{IC}_{50}$ values of $11.4 \mu \mathrm{M}$ for compound $\mathbf{2 , 9 . 3} \mu \mathrm{M}$ for compound $\mathbf{3}, 33.9$ $\mu \mathrm{M}$ for compound $\mathbf{4}$, and $11.1 \mu \mathrm{M}$ for compound 5.


## Experimental Section

General Experimental Procedures. UV and IR spectra were recorded on a Hewlett-Packard model 8451 A diode array spectrophotometer and a Nicolet Magna-IR model 550 spectrometer, respectively. NMR spectra ( $\delta \mathrm{ppm}$, J in Hz) were obtained on a Bruker AC200 spectrometer. HREIMS were determined with a VG-ZAB-SEQ instrument at 70 eV . HPLC separations were carried out using a Thermo Separations pump and UV detector, a Shodex RI-71 detector, and a YMC RP$18(20 \times 250 \mathrm{~mm})$ column.
Animal Material. The green tunicate Aplidium meridianum ${ }^{9,10}$ was collected by trawling at a depth of 100 m near the South Georgia Islands and stored at -20 ${ }^{\circ} \mathrm{C}$ until analyzed. Taxonomic classification was carried out by one of us (M. T.). Aplidium meridianum forms very soft, dome-shaped green colonies with a maximun diameter of 7 cm . A voucher specimen is deposited at the Cátedra de Anatomía Comparada, Facultad de Ciencias Exactas, Físicas y Naturales, Universidad Nacional de Córdoba, Córdoba, Argentina.
Extraction and Isolation. The frozen tunicate (900 g) was triturated and extracted three times with EtOH ( 2 L ). This extract was taken to dryness under reduced pressure to yield a yellow residue that was flash-
chromatographed on reversed-phase Si using an $\mathrm{H}_{2} \mathrm{O}-$ MeOH gradient. The fraction eluted with $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ (50:50) ( 1.5 g ) was chromatographed on Sephadex LH$20(4 \times 80 \mathrm{~cm}$ column, MeOH ) to yield meridianin A ( 26 mg ). The fraction eluted from reversed-phase with $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ (80:20) (1.7 g) was similarly chromatographed on Sephadex LH-20, and 24 fractions were collected. Fraction 23 gave pure meridianin B (2) (18.7 mg ); fractions 17 and 18 were pooled and further separated by HPLC using $\mathrm{H}_{2} \mathrm{O}-\mathrm{MeCN}(70: 30)$ as eluent to yield pure meridianins C (3) (12 mg) and D (4) (16 mg). Fraction 19 was purified by HPLC using MeOH $\mathrm{H}_{2} \mathrm{O}$ (78:22) to give pure meridianin E (5) ( 37 mg ).

Meridianin A (1): recrystallized from $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ as yellow needles; mp $164-168^{\circ} \mathrm{C}$; UV $\left(\mathrm{CH}_{3} \mathrm{Cl}\right) \lambda_{\text {max }}(\mathrm{log}$ є) 248 (3.68), 356 (3.58) nm; IR (KBr) $v_{\max } 3437,3351$, $3200,2924,1647,1605,1533,1469,1326,820,721 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR see Tables 1 and 2 , respectively; HREIMS m/z 226.0857 (calcd for $\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{4} \mathrm{O}, 226.0855$ ).

Meridianin B (2): recrystallized from EtOAc as a yellow powder; mp $190{ }^{\circ} \mathrm{C}(\mathrm{dec}) ;$ UV $\left(\mathrm{CH}_{3} \mathrm{Cl}\right) \lambda_{\text {max }}(\mathrm{log}$ є) 246 (3.87), 354 (3.71) nm; IR (KBr) $v_{\max } 3452,3357$, 3226, 2919, 1634, 1590, 1532, 1466, 1429, 1327, 1225, $816 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR see Tables 1 and 2, respectively; HREIMS m/z 303.9959 (calcd for $\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{~N}_{4}$ OBr, 303.9960).

Meridianin C (3): recrystallized from $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ as a yellow powder; mp $103-106{ }^{\circ} \mathrm{C}$; UV $\left(\mathrm{CH}_{3} \mathrm{Cl}\right) \lambda_{\text {max }}$ ( $\log \epsilon$ ) 244 (4.06), 324 (4.10) nm; IR (KBr) $\nu_{\max } 3399$, 3326, 3190, 2917, 1666, 1586, 1514, 1450, 1169, 880, 808, 784, $664 \mathrm{~cm}^{-1}{ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR see Tables 1 and 2, respectively; HREIMS m/z 288.0008 (cal cd for $\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{~N}_{4} \mathrm{Br}, 288.0011$ ).

Meridianin D (4): recrystallized from EtOAcMeOH as a yellow powder; mp $218-221^{\circ} \mathrm{C}$; UV $\left(\mathrm{CH}_{3}-\right.$ $\mathrm{Cl}) \lambda_{\max }(\log \epsilon) 240(4.17), 324(4.14) \mathrm{nm} ; \mathrm{IR}(\mathrm{KBr}) \nu_{\max }$ 3432, 3325, 3263, 3176, 2925, 1663, 1573, 1516, 1449, $891,818 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR see Tables 1 and 2, respectively; HREIMS m/z 288.0007 (cal cd for $\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{~N}_{4}-$ $\mathrm{Br}, 288.0011$ ).

Meridianin E (5): recrystallized from $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ as yellow crystals; mp $172-175{ }^{\circ} \mathrm{C}$; UV (MeOH) $\lambda_{\max }$ $(\log \epsilon) 224$ (4.20), 358 (3.85) nm; IR (KBr) $v_{\max } 3387$,

3335, 3226, 2927, 1634, 1590, 1538, 1392, 1225, 802, $721 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR see Tables 1 and 2, respectively; HREIMS m/z 303.9960 (calcd for $\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{~N}_{4}{ }^{-}$ OBr, 303.9960).

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